## Photochemical Synthesis of Bicyclo[4.1.0]hept-2-enes

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Bicyclo[4.1.0]hept-2-enes (norcarenes) were obtained in good yield by photolysis of 3-methylthiomethyl or 3-(2-ethoxy-carbonyl)vinylbicyclo[2.2.2]oct-5-en-2-ones in benzene upon high pressure Hg lamp irradiation.

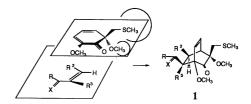
The photochemistry of the bicyclo[2.2.2]oct-5-en-2-ones (I) has been investigated widely. Although it is well-known that the bicyclo[4.2.0]octenones (II) are obtained via the 1,3-acyl shift,<sup>1</sup> the formation of the decarbonylation products, III were scarcely obserbed owing to their successive rapid photoisomerization,<sup>2</sup> except for a few cases.<sup>3</sup> We found that irradiation

of the bicyclo[2.2.2]octenones (1 or 2) introduced an appropriate group such as methylthiomethyl or (2-ethoxycarbonyl)vinyl group at C<sub>3</sub> position, gave the corresponding norcarenes, bicyclo[4.1.0]hept-2-enes (3 and 4).

Pyrex filter 
$$R^2$$
  $R^2$   $R^3$   $R^3$ 

The bicyclo[2.2.2]oct-5-en-2-ones (1a-j) were obtained by cycloaddition of cyclohexadiene (6)<sup>4</sup> with a variety of dienophiles (7a-j), in toluene in a sealed tube at 100 °C.<sup>5</sup> The

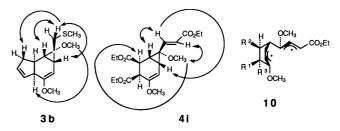
stereochemistry of the compounds was confirmed (to be the endo adducts) by <sup>1</sup>H-<sup>1</sup>H coupling constants or nuclear Overhauser effect (nOe). This is explained by the endo attack



Scheme 1.

of the dienophiles from the less hindered face (Scheme 1). 3-(2-Ethoxycarbonyl)vinylbicyclo[2.2.2]oct-5-en-2-ones (2) was prepared by Wittig reaction<sup>6</sup> of aldehyde (8) which was obtained via oxidation with sodium metaperiodate followed by Pummerer reaction with trifluoracetic anhydride<sup>7</sup> in 38-77% total yield from 1.8

A solution of 1 in benzene (1.4 x 10<sup>-2</sup> M) was irradiated with a 400-W high pressure Hg lamp for an appropriate time in a Pyrex immersion well at 0 °C. The solvent was evaporated, and the resiedue was purified by column chromatography on silica gel (100-200 mesh, Micro Bead 4B, Fuji Davison Chemical Ltd., eluent; 10% ethyl acetate in hexane) afforded the desired norcarenes (3) in moderate yields (Table 1). A short irradiation gave mainly the 1,3-acyl shift products (5) (entry 7), which was converted to the decarbonylation products (3) after longer reaction time (entry 8). Although in the photolysis of the compounds having acyl or formyl group, rearrangement of acyl group or deformylation occurred (entry 9-11), the desired norcarenes (3k-m) were obtained by protection of these acyl or formyl group as an acetal (entry 12-14). The structure of these norcarenes (3a-p) was confirmed by the nOe difference spectra



Scheme 2, NOes obserbed in 3 b and 4 s.

and the other spectra. For example, the nOe experiment of  $3\mathbf{b}$  showed the cis arrangement of 1-H, 6-H, and methylthiomethyl group at C-7 (Scheme 2). This suggested that the decarbonylation-recyclization sequence proceeded retaining the stereochemistry at the quarternary carbon (C-7 of 3). The <sup>1</sup>H NMR spectra displayed resonance at  $\delta$  1.0-1.4 ( $J_{1,6}$  9.2-9.6 Hz) for two characteristic protons (nOe) of cyclopropane ring junction<sup>9</sup> and at  $\delta$  4.7-4.9 as doublet for a olefinic proton of the

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## cyclohexene ring.

In direct irradiation of 2 in benzene upon Pyrex immersion well, the desired norcarenes (4) were obtained in much better yields than 3 (Table 2). This was due to the stability of the biradical intermediate (10) in the photolysis. In order to complete the reaction, 2a-f required only 2-5 h, but 2g-i needed some-what long irradiation time (10-12 h). The structure of 4 was confirmed in a similar manner as 3 (nOe of 4i was shown in Scheme 2).

Further exploitation of this transformation as a versatile approach to sesquiterpenes such as sesquicarene and sirenin is under study.

Table 1. Synthesis of 3 by photolysis of 1

Entry		1		Irradiation	Yield / %		
		R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	time / h	3	5
1	а	Q		Н	8	46	-
2	b	-CH=CH-CH <sub>2</sub> -		Н	8	59	-
3	С	C(CH <sub>3</sub> )=CH <sub>2</sub>	Н	Н	8	59	-
4	d	CO <sub>2</sub> Et	Н	Н	10	71	-
5	е	CO <sub>2</sub> tBu	Н	Н	4	69	8
6	f	CO <sub>2</sub> Me	Н	Ме	4	58	14
7	g	COEt	Н	Н	2	9	66
8	g	COEt	Н	Н	10	48	-
9	h	COMe	Н	Ме	5	26 ª	-
10	i	сно	Н	Н	5	0 ь	-
11	j	СНО	Н	Мe	5	0 с	-
12	k	C(OCH <sub>2</sub> ) <sub>2</sub> Me	Н	Н	5	56	-
13	1	CH(OCH <sub>2</sub> ) <sub>2</sub>	Н	Н	5	52	-
14	m	CH(OCH <sub>2</sub> ) <sub>2</sub>	Н	Ме	5	54	-

a The norcarene (3n; R<sup>3</sup>= CH<sub>3</sub>, R<sup>4</sup>= COCH<sub>3</sub>) was also formed by rearrangement of acetyl group in 35% yield.

b The norcarene (30;  $R^3$ ,  $R^4$  = H) was  $R^3$  formed after deformylation in 22% yield.

OCH<sub>3</sub>

3n-p

c The norcarene (3p;  $R^3$  = CH<sub>3</sub>,  $R^4$  = H) was formed after deformylation in 42% yield.

Table 2. Synthesis of 4 by photolysis of 2

Entry		2			Irradiation Yield / %		
		R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	time / h	4	2
1	а		•	Н	5	62	-
2	b	-CH=CH-CH <sub>2</sub> -		н	5	65	-
3	С	C(CH <sub>3</sub> )=CH <sub>2</sub>	Н	Н	5	60	-
4	d	CO <sub>2</sub> Et	Н	Н	5	92	-
5	е	CO <sub>2</sub> tBu	Н	Н	5	72	-
6	f	CO <sub>2</sub> Me	Н	Me	2	77	-
7	g	CN	Н	Н	10	56	14
8	h	Н	CO <sub>2</sub> Et	CO <sub>2</sub> Et	12	<b>7</b> 5	-
9	i	CO <sub>2</sub> Et	CO <sub>2</sub> Et	Н	12	72	-
10	j	CH(OCH <sub>2</sub> ) <sub>2</sub>	Н	Мe	5	64	-
11 k CH=CHCO <sub>2</sub> Et H			Ме	8	55	33	

## References and Notes

- V. K. Singh and M. Porinchu, J. Chem. Soc., Chem. Commun., 1993, 134.
- 2 K. N. Houk, Chem. Revs., 76, 1 (1976).
- 3 H. Hart, R. K. Murray, Jr., and G. D. Appleyard, Tetrahedron Lett., 1969, 4785; I. M. Takakis and W. C. Agosta, Tetrahedron Lett., 1978, 531.
- 4 S. Katayama, T. Watanabe, and M. Yamauchi, *Chem. Pharm. Bull.*, 41, 439 (1993).
- 5 1k was obtained by cycloaddition of 6 with 7k (R<sup>1</sup>= COCH<sub>3</sub>, R<sup>2</sup>= R<sup>3</sup>= H), followed by refluxing with ethylene glycol and p-TsOH; 11-m was derived from 1i-j, repectively, by treatment ethylene glycol and p-TsOH.
- 6 A. Balakumar, S. Janardhanam, and K. Rajagopalan, J. Org. Chem., 58, 5482 (1993).
- 7 H. Sugihara, R. Tanikaga, and A. Kaji, Synthesis, 1978, 881
- 8 In these Pummerer reaction process, 3-bis(methylthio)-methyl derivative (9) generated as a byproduct together with 8. These structures were confirmed by several spectral data such as <sup>1</sup>H NMR and conversion into 8 by treatment with AgClO<sub>4</sub>; T. Mukaiyama, S. Kobayashi, K. Kamio, and H. Takei, *Chem. Lett.*, **1972**, 237.
- 9 K. E. Harding, J. B. Strickland, and J. Pommerville, *J. Org. Chem.*, **53**, 4877 (1988).